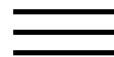


Generation of size-controlled palladium (0) and gold (0) nanoclusters inside the nanoporous domains of gel-type functional resins: Part II: Prospects for oxidation.

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Journal of Molecular Catalysis A: Chemical

Volume 238, Issues 1–2, 1 September 2005, Pages 26–34

Generation of size-controlled palladium(0) and gold(0) nanoclusters inside the nanoporous domains of gel-type functional resins: Part II: Prospects for oxidation catalysis in the liquid phase

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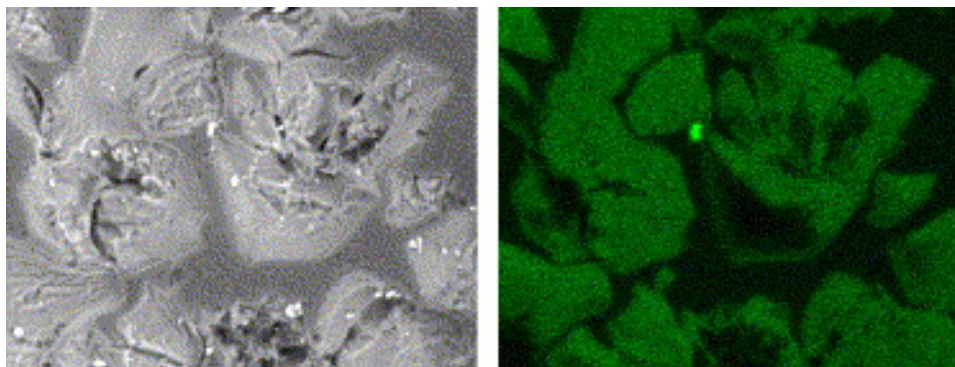
Abstract

Moderately cross-linked co-polymers of *N,N*-dimethylacrylamide (DMAA), 2-(methylthio)ethyl methacrylate (MTEMA) and *N,N*-methylenebisacrylamide proves to be effective macromolecular ligands able to extract, Pd<sup>II</sup> and Au<sup>III</sup> from water solutions and to thoroughly disperse them inside the relevant polymer frameworks. Chemical reduction with NaBH<sub>4</sub> in water leads to M<sup>0</sup>/resin composites, in which size-control of the generated metal nanoclusters is achieved. Catalysts Au<sup>0</sup>/MTEMA-DMAA are active in the rapid oxidation of *n*-butanal to *n*-butanoic acid by dioxygen under mild conditions

in water. Catalysts  $M^0$ /MTEMA-DMAA and  $M^0$ /C ( $M = Au, Pd$ ) are active and moderately chemoselective in the oxidation of *n*-butanol to *n*-butanal under the same conditions. Activity and chemoselectivity are positively affected by the co-presence of the two metal centres and for  $M = Au$  and reach the best level when the very hydrophilic resin poly-(4-vinylpyridine-acrylic acid-*N,N'*-methylenebisacrylamide) (VAM) is employed.

## Graphical abstract

Distribution of gold through a section of a MTEMA-DMAA 4–6 particle.



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## Keywords

$Pd^0$ ;  $Au^0$  nanoclusters; Functional resins; Oxidation catalysis

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