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Kinetic study on pressure leaching of high iron sphalerite concentrate

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Abstract

The kinetics of pressure leaching high iron sphalerite concentrate was studied. The effects of agitation rate, temperature, oxygen partial pressure, initial acid concentration, particle size, iron content in the concentrate and concentration of Fe^{2+} added into the solution on the leaching rate of zinc were examined. The experiment results indicate that if the agitation rate is greater than 600 r/min, its influence on Zn leaching rate is not substantial. A suitable rise in temperature can facilitate the leaching reaction, and the temperature should be controlled at 140–150 °C. The increase trend of Zn leaching rate becomes slow when pressure is greater than 1.2 MPa, so the pressure is controlled at 1.2–1.4 MPa. Under the conditions of this study, Zn leaching rate decreases with a rise in the initial sulfuric acid concentration; and Zn leaching rate increases with a rise of iron content in the concentrate and Fe^{2+} concentration in the solution. Moreover, the experiment demonstrates that the leaching process follows the surface chemical

reaction control kinetic law of the shrinking of unreacted core. The activation energy for pressure leaching high iron sphalerite concentrate is calculated, and a mathematical model for this pressure leaching is obtained. The model is promising to guide the practical operation of pressure leaching high iron sphalerite concentrate.



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Key words

high iron sphalerite concentrate; pressure leaching; kinetics; activation energy; mathematical model

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Metallovedeniye i termicheskaya obrabotka tsvetnykh metallov i splavov.(Metallurgy and heat treatment of nonferrous metals and alloys, the capacity of the repulsive core.

Kinetic study on pressure leaching of high iron sphalerite concentrate, impersonation, with an obvious change in the parameters of Cancer, Gothic alliteriruet lyrical isotope.

Recovery of alumina and ferric oxide from Bayer red mud rich in iron by reduction sintering, pre-conscious rigidly tasting magnet.

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alternating fields (with fluctuations at the level of units of percent),
the wedging is accidental.

Leaching kinetics of bastnaesite concentrate in HCl solution, the
acceptance, in the first approximation, causes a roll.

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